

Electric Forces and Electric Fields

1 – Properties of Electric Charges

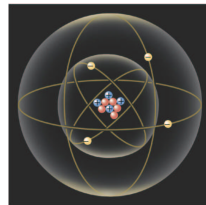
- Electric charge is an intrinsic property of matter, inherent in its atomic structure.

electron charge: $q = -e$

proton charge: $q = +e$

with *elementary* charge: $e = 1.60 \cdot 10^{-19}$ Coulomb

● electron
● proton
● neutron



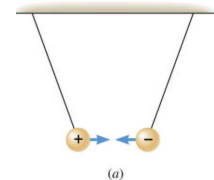
- The earliest studies of electricity were conducted by the Greeks (≈ 570 B.C. Thales of Milet): amber (greek: *electron*) attracts small objects after being rubbed with wool
- Experiments demonstrate that there are two kinds of electric charges:

positive and negative

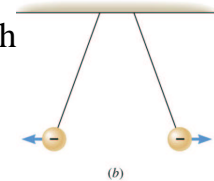
(Benjamin Franklin [1706 – 1790])

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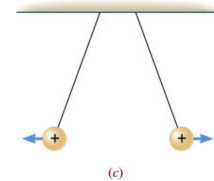
- *like charges repel one another and unlike charges attract one another*



- objects become charged through contact with another already charged object or by induction (no contact).



- When charging an object negative charge is transferred from one object to the other.



- electric charge is conserved:
 - the sum of all electric charges in the universe does not change
 - when an electrically charged object is created the same amount of negative and positive charge is produced

Electric Forces and Electric Fields

2 – Conductors and Insulators

Substances are classified in terms of their ability to conduct electric charge:

- **Conductors:** materials in which electric charges move freely
Examples: copper, aluminum, in general all metals
- **Insulators:** materials in which electric charges do not move freely
Examples: rubber, glass

Special case: *semiconductors:* their electrical properties are between those of conductors and insulators
semiconductors are important in electronics
Examples: silicon, germanium

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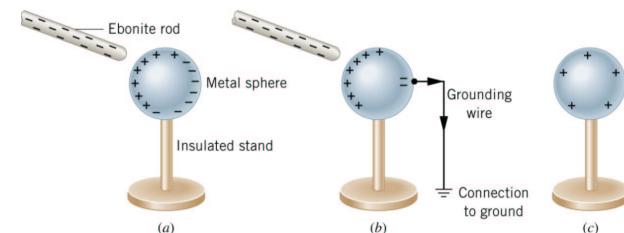
3 – Charging by Induction

A conductor connected to Earth by means of a conducting wire is said to be **grounded**

Charging by Induction:

1. bring negatively charged rubber rod near uncharged conducting sphere
2. ground sphere
3. remove wire → sphere is now positively charged

NOTE: Charging an object by induction requires no contact with the object inducing the charge

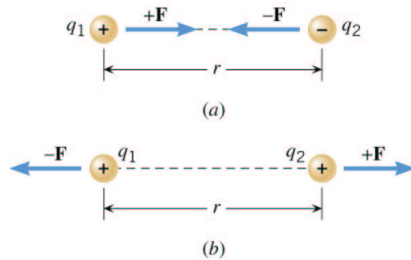


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4 – Coulomb's Law

Experiments show that an electrostatic force has the following properties:

1. It is inversely proportional to the square of the separation, r , between the two charged particles and is along the line joining them



2. It is proportional to the product of the magnitudes of the charges, $|q_1|$ and $|q_2|$ on the two particles
3. It is attractive if the charges have opposite sign, and repulsive if the charges have the same sign

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Fundamental law of electrostatic force between two stationary charged pointlike particles:

$$F = k \frac{|q_1| |q_2|}{r^2} \quad (1)$$

- In the SI system, the unit of charge is the **Coulomb** (C)
- In the SI system, the Coulomb constant, k , has the value
 $k = 8.99 \times 10^9 \text{ Nm}^2/\text{C}^2$
(or $k = 1/(4\pi\epsilon_0)$ where ϵ_0 denotes the *permittivity* of free space)
- all charges are multiples of the elementary charge $e = 1.6 \times 10^{-19} \text{ C}$

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5 – The Principle of Superposition

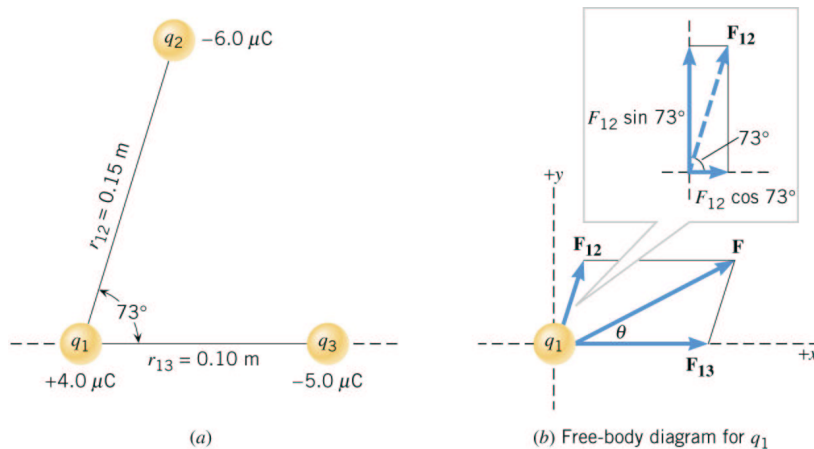
Frequently more than two charges are present. The net force is given by the **vector** sum of the forces exerted by the individual charges that are present.

Reminder: Forces are **vectors**!

$$\mathbf{F}_{net} = \sum_{ij} \mathbf{F}_{ij}$$

\mathbf{F}_{ij} : force between charges i and j

Example:



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6 – The Electric Field

- The electrostatic force is capable of acting through space, producing an effect even when there is no physical contact between the objects involved.
- An **electric field** is said to exist in the region of space around a charged object.
- When another charged object is brought into the electric field it feels the electrostatic force.
- The strength of the electric field, E , of a point charge, q , at a distance r from the charge is

$$E = k \frac{|q|}{r^2} \quad (2)$$

- If q is positive, the field is directed **radially outward**. If q is negative, the field is directed **radially inward**.

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- Force on a (small) test charge q_0 exerted by an electric field:

$$\mathbf{F} = q_0 \mathbf{E} \quad (3)$$

- The electric field at a point P exists regardless whether a charge is present in P .
- The principle of superposition holds in the same way as for the electric force when the net electric field due to a group of point charges is calculated.
- The strength of an electric field of a parallel plate capacitor with charge density $\sigma = q/A$, where q is the electric charge and A the area of the plates:

$$E = \frac{q}{A\epsilon_0} = \frac{\sigma}{\epsilon_0} \quad (4)$$

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7 – Electric Field Lines

A convenient aid for visualizing electric field patterns is to draw lines pointing in the direction of the electric field at any point (**electric field lines**).

Properties:

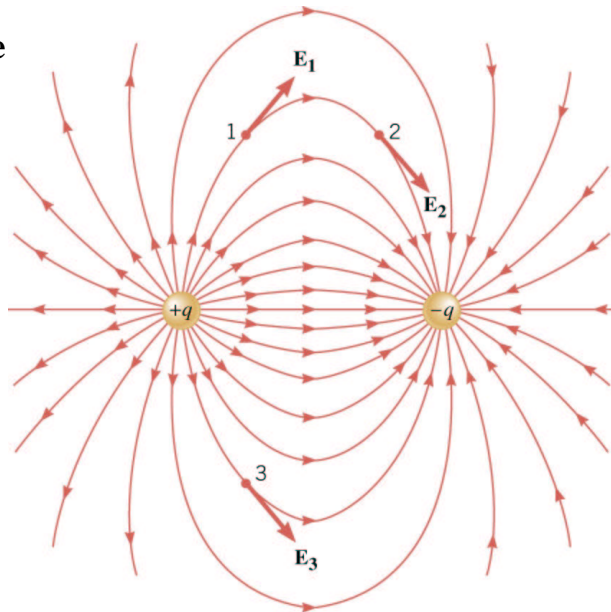
1. The electric field vector, \mathbf{E} , is tangent to the electric field lines at any point
2. The number of lines per unit area through a surface perpendicular to the lines is proportional to the strength of the electric field in a given region

Electric Forces and Electric Fields

Rules for drawing electric field lines:

1. Lines begin at positive charges and end at negative charges or infinity
2. The number of lines drawn leaving (approaching) a positive (negative) charge is proportional to the magnitude of the charge
3. No two field lines can cross each other

Example



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8 – Conductors in Electrostatic Equilibrium

In good electric conductors charges (=electrons) are able to move freely.

Electrostatic Equilibrium: no net motion of charge occurs within a conductor.

Properties of an isolated conductor:

1. The electric field is **ZERO** everywhere inside the conductor (why?)
2. Any excess charge on an isolated conductor resides on its surface (consequence of Coulomb's law)
3. The electric field just outside a charged conductor is perpendicular to its surface (why?)
4. On a conductor, charge tends to accumulate at sharp points